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Paperback – March 12, 2009 by James A. Vyvyan Donald L. Pavia, Gary M. Lampman, George S. Kriz (Author)

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Pavia, Lampman and Kriz CHAPTER 1

_____ 1. (a) 90.50% carbon; 9.50% hydrogen (b) C₄H₅ 2. 32.0% carbon;

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5.4% hydrogen; 62.8% chlorine; C₃H₆Cl₂. C₂H₅NO₂. 180.2 = molecular mass. Molecular formula is C₉H₈O₄. 5. Equivalent weight = 52.3 ...

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