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**A-level**

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in Chemistry A General  
Information . Molar  
-gas volume = 24.0  
dm<sup>3</sup> mol<sup>-1</sup> at room  
temperature and  
pressure, RTP

Avogadro constant,  $N_A$   
=  $6.02 \times 10^{23} \text{ mol}^{-1}$

Specific heat capacity  
of water,  $c_p = 4.18 \text{ J g}^{-1} \text{ K}^{-1}$

Ionc product of

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water,  $K_w = 1.00 \times 10^{-14} \text{ mol}^2 \text{ dm}^{-6}$  at 298 K  
1 tonne = 10<sup>6</sup> g  
Arrhenius equation:  $k = A e^{-E_a/RT}$  or  $\ln k = -$

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specifications will help you to inspire students, nurture their passion for the subject and lay the foundations for further study and the workplace. Our specifications allow you to choose the context and applications to bring chemistry to life in the way that best suits the needs of your students.

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questions.

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CHEMISTRY Paper 3 .

Do not write outside  
the box ... Use these  
data and your answer  
to Question 04.3 to plot  
a graph of  $V$  (y-axis)  
against  $t$  (x-axis) for  
mixture E. Draw a best-  
fit straight line through  
your points and

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calculate the gradient  
of this line.

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**A-level**

For first award of AS  
level in Summer 2017

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